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$O_2 + H_2 \rightarrow 2H_2O$  O H H O H H H H O O Now we can cancel off oxygen atoms. However, hydrogen atoms from reactant is short 2 atoms. Therefore, we multiply the hydrogen molecule by 2 so that the short is balanced. The equation is fully balanced when we are able to cancel off all atoms of that element on both sides.  $O_2 + 2H_2 \rightarrow 2H_2O$  O H H O H H H H H H O O 3.11 Calculations from Equations Reacting Masses ...

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